## THE SHERADSKY REARRANGEMENT OF $\alpha, \alpha$ -disubstituted cyclopentanone aryloximes : A synthesis of the sesquiterpenes (±)-aplysin and (±)-filiformin

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**Abstract** : Lactols <u>6a,b</u>, obtained by the Sheradsky rearrangement of aryloximes <u>o</u>, were alkylated in to <u>9a</u>, in which three contiguous chiral centers were controlled. Cyclization of <u>9a</u> gave the marine sesquiterpenes  $(\pm)$ -aplysin <u>11</u> and  $(\pm)$ -filiformin <u>13</u>.

The Sheradsky rearrangement of aryloximes  $\underline{o}^1$  (Scheme), a modification of the Fischer indole<sup>2</sup> rearrangement of phenylhydrazones <u>n</u>, has apparently never been studied when starting from an  $\alpha, \alpha'$ -disubstituted cycloalkanone. In both cases, enolisation, followed by rearrangement and cyclisation to a similar intermediate <u>a</u>, should give an NH<sub>2</sub>-substituted tricyclic compound <u>1</u>.



## Scheme

With cyclohexanone (n=3), the Fischer indole reaction finally yields an indolenine i, upon elimination of ammonia from t (X=NH). We have recently demonstrated the easy nitrous deamination of indolenines i to hemiketals  $\underline{h}^3$ , and we planned to extend the scheme to cyclopentanones (n=2), with the synthesis of benzofuranic sesquiterpenes in view. However, the Fischer "indolisation" of  $\alpha$ -substituted cyclopentanones failed to give indolenines  $\underline{i}^4$ . Therefore, we turned to the Sheradsky rearrangement of aryloximes  $\underline{o}$  (n=2), which could possibly lead to hemiketals  $\underline{h}$ .

5-Aminoxy-2-bromotoluene 1 was prepared by a rather lengthy known procedure<sup>5</sup> and further reacted with 2,5-dimethylcyclopentanone 2 to yield oximes 3a,b, along with notable quantities of 3-bromo-m-cresol<sup>6</sup>. <u>3a</u> and <u>3b</u> could be separated and characterised ; however, owing to their instability, the crude reaction mixture was used in the following reaction.



















9 b

<u>10a</u>

R=-COPh : 12 a R=-H : isolaurinterol

C-3- $\alpha$ Me : <u>11a</u> : aplysin C-3- BMe : 11b : epiaplysin

13 : filiformin

The acid catalysed rearrangement (TsOH, MeOH, reflux) furnished three groups of products : aminals  $4\underline{a},\underline{b}$  (24%), their regioisomers  $5\underline{a},\underline{b}$  (23%) and 3-bromo-m-cresol (25%). The cyclisation was completely devoid of regioselectivity and the a/b ratio in 4 and 5 was the same as in 3 (NMR). The mixture of the four aminals ( $4-5\underline{a},\underline{b}$ ) was heated in 60% aqueous acetic acid. Only  $4\underline{a},\underline{b}$  were converted into hemiketals  $6\underline{a},\underline{b}$ . In  $5\underline{a},\underline{b}$ , the formation of the intermediate oxonium was prevented as there would be a strong steric interaction of the two methyl groups at C-8 and C-8b. This allowed a very clean isolation of the non-basic hemiketals  $6\underline{a},\underline{b}$  from the reaction mixture.

Alkylation of the diastereoisomeric mixture <u>6a,b</u> with MeMgBr required a large excess of reagent, and prolonged reaction times. However, this reaction was highly rewarding as it fortuitously allowed a stereochemical control of the three adjacent asymmetric centres of the cyclopentane ring : in most runs, the pure (racemic)  $2a^7$  was the only product (60%) of the reaction (in some runs, 10% of its diastereomer at C-3 <u>9b</u> was obtained). This result (at least) implies epimerisation of the secondary methyl group at C-3 in <u>6b</u>, through the magnesium pheno-enolate <u>8</u>, which could be formed in the second step of the reaction from the quasi-rigid Mg<sup>++</sup>-chelating keto-phenolates <u>7a</u> and <u>7b</u>. In the case of <u>7b</u>, the carbonyl group is severely hindered, thus making enolisation (-8) faster than alkylation (-9b).

The role of metal cations in the stereoselectivity of reactions of lactols with organometallic or hydride reagents has been recently emphasised<sup>8</sup>. To the best of our knowledge, this had never been combined with an epimerisation of a vicinal chiral center.

The dibenzoate <u>10a</u>, when heated at 350°C under reduced pressure (0.1 mmHg), led in poor yield to a mixture of  $(\pm)$ -aplysin <u>11a</u> and benzoate <u>12a</u> ( $(\pm)$ -isolaurinterol benzoate)<sup>9</sup> but not to 3-epiaplysin <u>11b</u>. These results, which formally achieve a stereocontrolled synthesis of  $(\pm)$ -aplysin, unambiguously prove the  $\alpha$ -configuration at C-3 in the starting pheno-alcohol <u>9a</u>.

Heating alcohol <u>9a</u> in acetic acid containing traces of TsOH gave ( $\pm$ )-aplysin <u>11a</u> (21%) and a mixture (32%) of ( $\pm$ )-filiformin <u>13</u> and ( $\pm$ )3-epiaplysin <u>11b</u> (3/1, NMR). Repetitive chromatography enabled us to obtain an analytical sample of pure <u>13</u> and <u>11b</u>. All IR, MS,<sup>1</sup>H and <sup>13</sup>C NMR data for <u>11a</u>, <u>11b</u> and <u>13</u> were identical with those described in the literature<sup>10</sup>.

These results compare favourably with previous syntheses of  $aplysin^{11a-e}$  and of filiformin<sup>11d</sup>.

Work is in progress in our laboratories on the synthesis of debromo-analogues, using the more expeditive synthesis of aryloximes  $\underline{o}$  recently developed by Alemagna and coworkers<sup>12</sup>.

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## Notes and references :

- Sheradsky T., Tetrahedron Lett., 1966, 5225.
  Robinson B., "The Fischer Indole Synthesis", J.Wiley, New-York, 1982.
- 3. Lévy J. and Sigaut F., Tetrahedron Lett., 1983, 24, 4983. Lévy J. and Sigaut F., Tetrahedron Lett., 1983, 24, 4987. Laronze J.Y., Laronze J., Patigny D. and Lévy J., Tetrahedron Lett., 1986, 27, 489.
- 4. These results will be discussed elsewhere.
- 5. Sheradsky T., Salemnick G. and Nir Z., Tetrahedron, 1972, 28, 3833.
- 6. Castellino A.J. and Rapoport H., J.Org.Chem., 1984, 49, 4399.
- 7. The I.R. spectrum of 9a showed internal H bonding, and its <sup>1</sup>H NMR spectrum exhibited a low field signal at 2.9 ppm (ddd); COSY experiments showed it to be C-1- $\alpha$ H, whose position in the plane of the aromatic ring is in favour of the stereochemistry depicted.
- Kraus G.A., Molina M.T. and Walling J.A., J.Chem.Soc., Chem.Commun., 1986, 1568. Tomooka K., Okinaga T., Suzuki K. and Tsuchihashi G.I., Tetrahedron Lett., 1987, 28, 6335. Tomooka K., Matsuzawa K., Suzuki K. and Tsuchihashi G.I., Tetrahedron Lett., 1987, 28, 6339.
  - Canonne P., Plamondon J. and Akssira M., Tetrahedron, 1988, 44, 2903.
- 9. 12a could not be completely characterised. However its <sup>1</sup>H NMR spectrum showed the two coupling olefinic signals at 4.95 (d) and 5.12 ppm (d, J=3Hz) of an O-acyl isolaurinterol derivative : Irie T., Suzuki M., Kurosav E. and Masamune T., Tetrahedron, 1970, 26, 3271. All other compounds were fully characterised (UV, IR, MS, <sup>1</sup>H and <sup>13</sup>C NMR).
- 10. Aplysin 11a : mp, IR, MS, <sup>1</sup>H NMR : Yamamura S. and Hirati Y., Tetrahedron, 1963, <u>19</u>, 1485. <sup>13</sup>C NMR : Sims J.J., Rose A.F. and Izac R.R. in "Marine Natural Products ; Chemical and Biological Perspectives", Scheuer P.J. Ed., Academic Press 1978, vol. 2, chap. 5. Filiformin 13: Kaslauskas R., Murphy R.J., Ronald J.Q. and Robert J.W., Aust.J.Chem., 1976, 29, 2533.

3-Epiaplysin 11b : our <sup>1</sup>H and <sup>13</sup>C NMR data are as follows (CDCl<sub>3</sub>) :



In his total synthesis of aplysin<sup>11a</sup>, Yamada obtained a by-product that he refered to an isomer of either aplysin or of filiformin. According to the published data we think it is 3-epiaplysin.

- 11.a) Yamada K., Yazawa H., Vemura D., Toda M. and Hirata Y., Tetrahedron, 1969, 25, 3509.
  - b) Ronald R.C., Tetrahedron Lett., 1976, 4413.
  - c) Ronald R.C., Gewaldi M.B. and Ronald P.B., J.Org.Chem., 1980, 45, 2224.
  - d) Goldsmith D.J., John T.K., Kwong C.D. and Painter III G.R., J.Org.Chem., 1980, 45, 3989.
  - c) Ghosh A., Biswas S. and Venkateswaran R.V., J.Chem.Soc., Chem.Commun., 1988, 1421.
- 12. Alemagna A., Baldolic C., Del Buttero P., Licandro E. and Maidrana S., Synthesis, 1987, 192. Baldoli C., Del Buttero P., Licandrro E. and Maidrana S., Synthesis, 1988, 344.

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